

Additions and Corrections

Transformation of a Methyleneamide Ligand at Molybdenum: Electrochemical Oxidation to a Cyanide, Reactions with Elemental Oxygen, Sulphur or Selenium and X-Ray Crystal Structures of *trans*-[Mo(CN)Cl(dppe)₂].MeOH and *trans*-[Mo(NCS)Cl(dppe)₂]; Electroreduction of the Cyanide to an Aminocarbyne, *trans*-[Mo(CNH₂)Cl(dppe)₂] (dppe = Ph₂PCH₂CH₂PPh₂) (1991, 121)

Adrian Hills, David L. Hughes, Colin J. Macdonald, Modher Y. Mohammed and Christopher J. Pickett

Page 126. In Scheme 3 the compound Cl-M'≡N-NH₂ should read Cl-M'≡C-NH₂

Nuclear and Electronic Contributions to the Photoreactivity of Iso- and Hetero-polyoxomolybdates (1991, 2063)

Bela Kraut and Guillermo Ferraudi

Page 2064. The wavelength scale of the top spectrum in Fig. 3 was inadvertently omitted and should range from 500 to 760 nm. Also, right-hand column, second paragraph. The value of λ_{max} for the ligand-radical polymolybdate I should read 550 nm.

Oxidation of Water by MnO₄⁻ mediated by Thermally Activated Ruthenium Dioxide Hydrate (1992, 1059)

Andrew Mills, Peter Douglas and Tom Russell

Page 1060. Equations (11) and (13) should be written as follows:

$$i_{\text{mix},t} = i_{001}^{r_2} i_{002}^{r_1} [\text{Red}^1]^{0_1 r_2} [\text{Ox}^2]^{0_2 r_1} \exp[\alpha_2 r_1 F(E_2^\circ - E_1^\circ)/RT] \quad (11)$$

$$i_{\text{mix},t} = K_1 i_{001}^{r_2} i_{002}^{r_1} [\text{Ox}^2]^{0_2 r_1} \exp[\alpha_2 r_1 F(E_2^\circ - E_1^\circ)/RT] \quad (13)$$

where $K_1 = [\text{H}_2\text{O}]^{0_1 r_2}$.

Page 1063, left-hand column, line 3 above equation (21). The term $[\text{Ox}^2]_{0_2 r_1}$ should be replaced by $[\text{Ox}^2]^{0_2 r_1}$.